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UDC 547.812.5'813.07

We have found that when phenylacetic acid and its 4-nitro derivative are heated in polyphosphoric ester (PPE), they undergo fourfold self-acylation to give compounds of the Ia,b type, which are converted to 2-hydroxy- γ -pyrones IIa,b by alkaline hydrolysis.

4 ArCH₂COOH
$$\frac{\text{ppE}}{\text{Ar}} \xrightarrow{\text{CH}_2} \text{OO} \xrightarrow{\text{O}} \text{OH} \xrightarrow{\text{Ar}} \text{CH}_2 \xrightarrow{\text{O}} \text{OH}$$

$$\text{I a,b} \xrightarrow{\text{CH}_2} \text{II a,b}$$

$$\text{I, II a Ar} = \text{C}_6\text{H}_5; \text{ b Ar} = 4\text{-NO}_2\text{C}_6\text{H}_5$$

A 3-g sample of phenylacetic acid was heated with vigorous stirring at 100° for 30 min in a tenfold excess of PPE, after which the mixture was hydrolyzed with cold water, heated carefully to the boiling point, and refluxed for 3-4 min. The hot aqueous solution of unchanged phenylacetic acid was decanted, and the residual oil began to crystallize on treatment with cold water to give colorless crystals of 2-phenylacetoxy-3,5-diphenyl-6-benzyl- γ -pyrone (Ia), with mp 120° (from alcohol), in 72% yield (based on the converted acid). PMR spectrum: s, 3.18 (2H); s, 3.28 (2H); m, 6.55-7.38 ppm (2OH). IR spectrum: 1755, 1640, 1550, and 1535 cm⁻¹.

When Ia was refluxed for 15 min in 30% aqueous KOH solution, red crystals of the known 2-hydroxy-3,5-diphenyl-6-benzyl- γ -pyrone (IIa), with mp 162-164° (from benzene), were formed.

The following compounds were similarly obtained after reaction at 100° for 50 min: 2- $(4-\text{nitrophenylacetoxy})-3,5-\text{di}(4-\text{nitrophenyl})-6-(4-\text{nitrobenzyl})-\gamma-\text{pyrone}$ (Ib) as light-brown crystals with mp 96-98° (from methanol) in 80% yield; 2-hydroxy-3,5-di(4-nitrophenyl)-6-(4-nitrobenzyl)- γ -pyrone (IIb) as light-brown crystals with mp 198° (from nitromethane). The character of the spectra of Ib and IIb is very similar to the character of the spectra of Ia and IIa. The results of analysis for C and H (and for N in the case of Ib and IIb) were in agreement with the calculated values for all of the compounds.

o-Bromo- and o-iodohomoveratric acids do not react under the described conditions, and homoveratric and homopiperonylic acids form different substances, the structure of which has not been ascertained.

Rostov State University. Scientific-Research Institute of Physical and Organic Chemistry, Rostov-on-Don 344006. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 5, pp. 705-706, May, 1977. Original article submitted December 14, 1976.

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